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INFLUENCE OF Ag DOPING ON THE MORPHOLOGICAL AND MAGNETIC PROPERTIES IN CuO NANOPOWDERS

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We studied the influence of Ag doping on the crystal structure and magnetic properties of CuO nanopowders. For the synthesis of nanoparticles of copper-silver oxides solid solutions with the composition $Cu_{1-x}Ag_xO$ (x=0.01–0.05), a selfpropagating synthesis was applied at room temperature, during which a successful reaction between metal nitrate and sodium hydroxide occurred. Prepared powders were calcinated at 700 °C for 2 h. The diffraction pattern was recorded at room temperature and atmospheric pressure without of any re-heating of the sample. The Rietveld method for fitting refinement procedure was performed which showed the incorporation of Ag³⁺ ions in the CuO crystal lattice, where they substitute Cu²⁺ ions. SQUID magnetometer was used for investigation of magnetic behavior of synthesized materials in temperature interval 2–400 K. It is known that copper(II) oxide exhibits ferroelectricity driven by magnetic order at temperature as high as 230 K [1]. Multiferroic phase is present above the first order phase transition at T_{NI} = 213 K and exists up to the subsequent first order phase transition T_{N2} = 230 K [1,2]. It was shown that disorder in the form of impurities can stabilize the ferroelectric phase [2] this was our motivation to dope CuO with Ag in order to improve further its multiferroic properties. Compared with CuO, we observed small changes in magnetic properties in Cu_{1-x}Ag_xO. The size and morphology of the particles were successfully determined by transmission electron microscopy (TEM) and scanning electron microscopy (SEM).

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